

- 1) A laser makes a beam with 0.1 watts of light at a wavelength of 630 nm. How many photons per second are in the beam?

$$0.1 \text{ watt} = (0.1 \text{ J/s}) \frac{(1 \text{ eV})}{1.602 \times 10^{-19} \text{ J}} = 6.24 \times 10^{17} \text{ eV/s}$$

$$E = \frac{hc}{\lambda} = \frac{1240 \text{ eV} \cdot \text{nm}}{630 \text{ nm}} = 1.968 \text{ eV each}$$

$$\text{Rate} = (6.24 \times 10^{17} \text{ eV/s}) / (1.968 \text{ eV/photon}) = 3.17 \times 10^{17} / \text{s}$$

$$3.17 \times 10^{17} / \text{sec.}$$

- 2) Muons at rest in the laboratory have a mean lifetime of 2.2×10^{-6} sec. How far, on the average, will these muons travel before decaying if they have a velocity of $0.6c$?

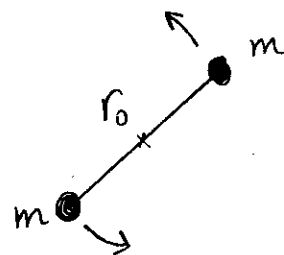
$$\gamma = \frac{1}{\sqrt{1-(0.6)^2}} = 1.25$$

$$\tau = \gamma \tau_0 = (1.25)(2.2 \times 10^{-6} \text{ s}) = 2.75 \times 10^{-6} \text{ s}$$

$$d = v \cdot t = (0.6)(3 \times 10^8 \text{ m/s})(2.75 \times 10^{-6} \text{ s})$$
$$= 495 \text{ m}$$

$$495 \text{ m}$$

3) The rotational energy of a molecule can be written as $E = L^2/2I$ where L is the angular momentum and I is the moment of inertia. Assuming that the angular momentum is quantized according to the Bohr's rule, $L = n\hbar$, find the wavelength of the photons emitted in the $n = 2 \rightarrow n = 1$ transition of the H_2 molecule. The moment of inertia for this molecule would be $I = \frac{1}{2}mr_0^2$ where $m = 938 \text{ MeV}/c^2$ and $r_0 = 0.074 \text{ nm}$.

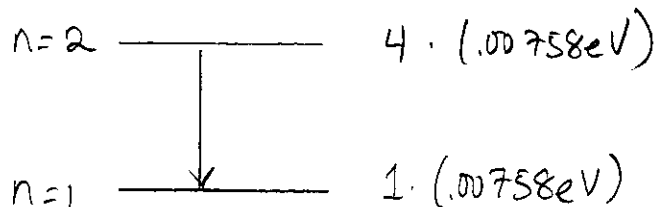


$$E = \frac{L^2}{2I} = \frac{(n\hbar)^2}{2(\frac{1}{2}mr_0^2)} = n^2 \frac{\hbar^2}{mr_0^2} = n^2 \frac{(\hbar c)^2}{r_0^2 mc^2}$$

$$= n^2 \frac{(1240 \text{ eV}\cdot\text{nm} / 2\pi)^2}{(0.074 \text{ nm})^2 (938 \times 10^6 \text{ eV})} = n^2 \cdot (.00758 \text{ eV})$$

So the photon energy is

$$E_{\text{photon}} = 3 \cdot (.00758 \text{ eV}) \\ = .02275 \text{ eV}$$



$$E = \frac{hc}{\lambda} \quad \lambda = \frac{hc}{E} = \frac{1240 \text{ eV}\cdot\text{nm}}{.02275 \text{ eV}} = 54511 \text{ nm}$$

54511 nm

- 4) If space had only two dimensions instead of three, the speed distribution of the atoms in a gas would be

$$n(v) = C v e^{-mv^2/2kT}$$

Find the average kinetic energy of the atoms for this distribution. [Hint: Use the normalization condition to find C .] The following integral will be useful: $\int_0^\infty x^{2n+1} e^{-ax^2} dx = \frac{n!}{2a^{n+1}}$

Normalize

$$\int_0^\infty n(v) dv = N = C \int_0^\infty v e^{-mv^2/2kT}$$

$$a = \frac{m}{2kT} \quad n=0$$

$$N = C \frac{0!}{2a} = \frac{C}{2a}$$

$$\underline{C = 2aN}$$

Now find average KE

$$\bar{E} = \frac{1}{N} \int_0^\infty \left(\frac{1}{2}mv^2\right) n(v) dv = \frac{1}{N} C \int \frac{1}{2}mv^3 e^{-mv^2/2kT} dv$$

$$= (2a) \left(\frac{1}{2}m\right) \int_0^\infty v^3 e^{-mv^2/2kT}$$

$$n=1, a = \frac{m}{2kT}$$

$$E = (a \cdot m) \frac{1!}{2a^2} = \frac{m}{2a} = \frac{m}{2} \frac{2kT}{m} = kT$$

kT

5) A π meson (rest energy 140 MeV) is traveling in the $+x$ direction with a kinetic energy of 80 MeV.

(a) Find the momentum of the π in MeV/c.

$$E^2 = (pc)^2 + (m_0c^2)^2$$

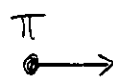
$$E = 140 \text{ MeV} + 80 \text{ MeV}$$

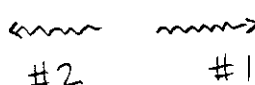
$$(pc)^2 = (220 \text{ MeV})^2 - (140 \text{ MeV})^2$$

$$pc = 169.7 \text{ MeV}$$

$$\boxed{169.7 \text{ MeV}/c}$$

(b) Suppose the π disintegrates, converting into two photons. Find the photon energies for the situation in which photon #1 is emitted in the $+x$ direction and #2 in the $-x$ direction.

BEFORE: 

AFTER: 

Energy conservation

$$E_\pi = E_1 + E_2$$

Momentum "

$$P_\pi = P_1 - P_2 \Rightarrow P_\pi c = P_1 c - P_2 c$$

For photons

$$pc = E \quad (\text{since } m=0)$$

so

$$P_\pi c = E_1 - E_2$$

$$E_1 + E_2 = E_\pi$$

$$E_1 - E_2 = P_\pi c$$

Add

$$2E_1 = E_\pi + P_\pi c$$

$$= (220 + 169.7) \text{ MeV}$$

$$E_1 = 194.85 \text{ MeV}$$

$$E_2 = (220 - 194.85) = 25.15 \text{ MeV}$$

$$E_1: \boxed{194.85 \text{ MeV}}$$

$$E_2: \boxed{25.15 \text{ MeV}}$$