

Chapter 27

P27.11 We use $I = nqAv_d n$ is the number of charge carriers per unit volume, and is identical to the number of atoms per unit volume. We assume a contribution of 1 free electron per atom in the relationship above. For aluminum, which has a molar mass of 27, we know that Avogadro's number of atoms, N_A , has a mass of 27.0 g. Thus, the mass per atom is

$$\frac{27.0 \text{ g}}{N_A} = \frac{27.0 \text{ g}}{6.02 \times 10^{23}} = 4.49 \times 10^{-23} \text{ g/atom}.$$

$$\text{Thus, } n = \frac{\text{density of aluminum}}{\text{mass per atom}} = \frac{2.70 \text{ g/cm}^3}{4.49 \times 10^{-23} \text{ g/atom}}$$

$$n = 6.02 \times 10^{22} \text{ atoms/cm}^3 = 6.02 \times 10^{28} \text{ atoms/m}^3.$$

Therefore,

$$v_d = \frac{I}{nqA} = \frac{5.00 \text{ A}}{(6.02 \times 10^{28} \text{ m}^{-3})(1.60 \times 10^{-19} \text{ C})(4.00 \times 10^{-6} \text{ m}^2)} = 1.30 \times 10^{-4} \text{ m/s}$$

$$\text{or, } v_d = \boxed{0.130 \text{ mm/s}}.$$

P27.14 (a) Applying its definition, we find the resistance of the rod,

$$R = \frac{\Delta V}{I} = \frac{15.0 \text{ V}}{4.00 \times 10^{-3} \text{ A}} = 3750 \Omega = \boxed{3.75 \text{ k}\Omega}.$$

(b) The length of the rod is determined from the definition of resistivity: $R = \frac{\rho \ell}{A}$. Solving for ℓ and substituting numerical values for R , A , and the value of ρ given for carbon in Table 27.1, we obtain

$$\ell = \frac{RA}{\rho} = \frac{(3.75 \times 10^3 \Omega)(5.00 \times 10^{-6} \text{ m}^2)}{(3.50 \times 10^{-5} \Omega \cdot \text{m})} = \boxed{536 \text{ m}}.$$

$$\text{P27.16 } J = \frac{I}{\pi r^2} = \sigma E = \frac{3.00 \text{ A}}{\pi(0.0120 \text{ m})^2} = \sigma(120 \text{ N/C})$$

$$\sigma = 55.3(\Omega \cdot \text{m})^{-1} \quad \rho = \frac{1}{\sigma} = \boxed{0.0181 \Omega \cdot \text{m}}$$

P27.42

$$R = \frac{(110 \text{ V})^2}{(500 \text{ W})} = 24.2 \ \Omega$$

$$(a) \quad R = \frac{\rho}{A} \ell \quad \text{so} \quad \ell = \frac{RA}{\rho} = \frac{(24.2 \ \Omega)\pi(2.50 \times 10^{-4} \text{ m})^2}{1.50 \times 10^{-6} \ \Omega \cdot \text{m}} = \boxed{3.17 \text{ m}}$$

$$(b) \quad R = R_0[1 + \alpha\Delta T] = 24.2 \ \Omega [1 + (0.400 \times 10^{-3})(1180)] = 35.6 \ \Omega$$

P27.43
$$R = \frac{\rho\ell}{A} = \frac{(1.50 \times 10^{-6} \ \Omega \cdot \text{m})25.0 \text{ m}}{\pi(0.200 \times 10^{-3} \text{ m})^2} = 298 \ \Omega$$

$$\Delta V = IR = (0.500 \text{ A})(298 \ \Omega) = 149 \text{ V}$$

$$(a) \quad E = \frac{\Delta V}{\ell} = \frac{149 \text{ V}}{25.0 \text{ m}} = \boxed{5.97 \text{ V/m}}$$

(b)

$$(c) \quad R = R_0[1 + \alpha(T - T_0)] = 298 \ \Omega [1 + (0.400 \times 10^{-3}/^\circ\text{C})320^\circ\text{C}] = 337 \ \Omega$$

P27.44 (a)
$$\Delta U = q(\Delta V) = It(\Delta V) = (55.0 \text{ A} \cdot \text{h})(12.0 \text{ V}) \left(\frac{1 \text{ C}}{1 \text{ A} \cdot \text{s}} \right) \left(\frac{1 \text{ J}}{1 \text{ V} \cdot \text{C}} \right) \left(\frac{1 \text{ W} \cdot \text{s}}{1 \text{ J}} \right) = 660 \text{ W} \cdot \text{h} = \boxed{0.660 \text{ kWh}}$$

$$(b) \quad \text{Cost} = 0.660 \text{ kWh} \left(\frac{\$0.0600}{1 \text{ kWh}} \right) = \boxed{3.96\text{¢}}$$